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Distribution of Σ 3 misorientations in polycrystalline strontium titanate

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Abstract

SrTiO₃ is a potential electrolyte material for solid-oxide fuel cells due to its high ion conductivity. Grain boundaries (GBs) play an important role in bulk ion conductivity. It has been reported that the resistance of $\Sigma 3$ GBs is much lower than the general GB in SrTiO₃. In order to clarify the conflicting reports on the prevalence of $\Sigma 3$ GBs in SrTiO₃, grain size and GB misorientations in Nb-doped and undoped SrTiO₃ have been investigated as a function of annealing time. The observations suggest that the prevalence of both low-angle GBs and $\Sigma 3$ GBs is strongly correlated to both abnormal grain growth and annealing time. In particular, the $\Sigma 3$ GB population approaches that predicted by the Mackenzie random distribution when abnormal grain growth dominates.

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1. Introduction

Strontium titanate (SrTiO₃) is an important perovskitestructured material, having superior electronic properties for various applications, including for solid-oxide fuel cells (SOFC)^{1,2} and grain boundary barrier layer capacitors.^{3,4} For SOFC, due to its high ion conductivity and good thermal stability at intermediate temperatures (600-800 °C), SrTiO₃ has been identified as a potential electrolyte material.^{1,2} In the SOFC electrolytes, the ion conductivity of SrTiO₃ is attributed to vacancy migration,^{1,2} which relates to GB properties. The main reason is that the GBs impede the transport of charge, and the resistance of the interface is attributed to the existence of a surface electrical potential.⁵ The electrical potential barrier is related to the individual defect formation energies, dopant concentration, temperature, oxygen pressure, and the dominant charge compensation mechanism in the bulk.⁶ Since the individual formation energies of defects such as cation and anion vacancies can differ, there should be an excess of the species with lower formation energy at a grain boundary at equilibrium. The GB becomes non-stoichiometric with a net charge. That net charge is

compensated with space charge regions near the grain boundary that maintain global electro-neutrality. The GB can be modelled as a double Schottky barrier and its excess resistance related to the charge carrier concentrations next to the GB core and in the bulk. The local equilibrium of defects with a GB depends on the character of the GB and can vary from GB to GB in a microstructure.

Since special properties (e.g. low GB energy) of coincidencesite lattice (CSL) GBs were proposed in polycrystalline materials,^{7,8} CSL GBs have been used to correlate with the electronic properties in perovskites. For example, in barium titanate (BaTiO₃), Σ 3 GBs having no positive temperature effect are reported to be PTC inert.^{9,10} In SrTiO₃, to understand how the Σ 3 GBs contribute to the electronic properties, the atomic structure and energy of Σ 3 GBs have been examined by various research groups. Initially, using high resolution transmission electron microscopy, Kienzle et al.¹¹ and Ernst et al.¹² characterized the atomic structure in bicrystals and polycrystals of SrTiO₃, respectively. Later, using density functional calculations, Hutt et al.¹³ calculated the electronic structure of the Σ 3 GBs based on Kienzle's¹¹ observations, and concluded that there is no evidence to support the existence of space charge layers at Σ 3 GBs. This result suggests that the resistance of Σ 3 GBs is much lower than the general GBs because no space charge layer impedes ion migration. So, an increased population of Σ 3 GBs results in a higher ion conductivity in SrTiO₃.

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The bulk ion conductivity is expected to depend on the number distribution of Σ 3 GBs. However, there are conflicting reports about the distribution of $\Sigma 3$ misorientations in SrTiO₃. Ernst et al.¹⁴ reported that in titanate-based ceramics (SrTiO₃, BaTiO₃ and Pb(Zr_xTi_{1-x})O₃) GBs with Σ 3 misorientations are found significantly more frequently than expected from a random misorientation distribution. However, Saylor et al.¹⁵ reported that the GBs with Σ 3 misorientations occur with a random distribution in undoped SrTiO₃ and they suggested that the result of Ernst et al.¹⁴ may be due to Fe-doping and TiO₂ excess. Seaton and Leach investigated the relationship between the microstructure and the populations of low Σ GBs for donor-doped (Pb, Sr, and Ca) BaTiO₃ during the early stages of sintering (less than 1 h).^{16,17} They found that the proportion of low Σ GBs increased during the first 30 min and then became stable. Recently Park et al.¹⁸ investigated the low Σ misorientations in undoped SrTiO₃ as a function of annealing time (1 and 16 h) and came to a different conclusion. They found that the proportion of Σ 3 misorientations decreased with annealing time.

In order to clarify whether preferred $\Sigma 3$ misorientations develop in SrTiO₃, this study investigates (i) grain size distributions, (ii) grain misorientation distributions and (iii) $\Sigma 3$ GB distributions in undoped and Nb-doped SrTiO₃ in samples annealed for different times. These three distributions have been determined by electron backscatter diffraction (EBSD).

2. Experimental

2.1. Ceramic synthesis

Samples were provided within the INCEMS Project by Prof. Hoffmann and Dr Bäurer (University of Karlsruhe). For a control experiment, undoped SrTiO₃ samples were prepared from SrCO₃ (99.9+ wt%, Sigma–Aldrich, Milwaukee, United States) and TiO₂ (99.9+ wt% Sigma-Aldrich, Milwaukee, United States) powders by the mixed-oxide route. The mixture of 0.996SrCO₃ to 1TiO₂ (molar ratio) was ball-milled for 4 h with 2-propanol in an attrition mill using ZrO₂ balls, calcinated at 975 °C in air for 6 h and then ball-milled for 16 h in a planetary mill. The calcined powder was uniaxially pressed into discs, isostatically densified at 400 MPa, sintered at 1425 °C in oxygen for 20 h and then quenched to room temperature. Nb-doped SrTiO₃ powder was prepared from Nb2O5 (99.9 wt%, ChemPur, Karlsruhe, Germany) and the same SrCO₃ and TiO₂ powders. The powder mixture of 0.994SrCO₃, 0.006Nb₂O₅ and 0.988TiO₂ (molar ratio) was processed by the same ball-milling and calcination procedures, and annealed at 1425 °C in oxygen for 1, 2, 4 or 20 h, and then quenched to room temperature. Dilatometry showed that samples attained greater than 97% of the theoretical density after short times at temperature.¹⁹

2.2. Grain size and misorientation

Grain size and grain misorientation analyses were carried out using EBSD. In order to obtain precise misorientation data, it is important to collect good quality Kikuchi maps



Fig. 1. The 3-D grain size (D_{3-D}) distribution in Nb-doped SrTiO₃ sintered for 1 h at 1425 °C in oxygen. Error bars indicate the standard deviation of grain size measurement from different sets measured on three regions that contain at least 600 grains each.

from EBSD, and this requires a well polished and uncontaminated sample surface.²⁰ The EBSD samples were prepared by mechanical grinding with SiC abrasive papers (800, 1200, 2000 mesh) and then polishing with a succession of 9, 3, 1 and 0.1 μ m diamond cloths. The samples were chemically etched in a 95H₂O–4HCl–1HF (by volume percent) solution for 30 s to remove surface contamination caused by sample polishing.

EBSD Kikuchi patterns were obtained using a JSM 6300F SEM, with an 80 μ A beam current and 0.19–4.64 μ m step size at a working distance of 15 mm. The step size was at least ten times smaller than the average grain size (diameter), to give precise measurements. The sample was inclined at 70° to the 20 kV incident electron beam. Analysis of the Kikuchi patterns was carried out with computer software (Inca, Oxford Instruments, UK). For grain determination, two neighbouring pixels were considered to be in the same grain if they had a misorientation angle less than 3°, and GBs were identified between adjacent grains.¹⁸ The grain boundary distribution statistics were based on the number of grain boundary segments identified between misorientated pixels.

In EBSD, the most common way to obtain the twodimensional (2-D) grain size distribution is by using the equivalent circle diameter.^{21,22} However, a description of grains (i.e. grain size) in three-dimensions (3-D) gives more physical insight than that in 2-D. In this study the 3-D grain size distribution was extracted from the equivalent circle diameters using the method developed by Underwood.²³ Error bars indicate the standard deviation of grain size measurement from different sets measured on three regions that each contain at least 600 grains. Bi-modal distributions of 3-D grain size are obtained, as shown in Fig. 1. This bi-modal distribution associates with abnormal grain growth, which is frequently found in SrTiO₃.^{24,25} The overall grain size (*G*) of a sample was calculated from $G = \bar{D}_{3-D,l} \times N_l/(N_l + N_s) + \bar{D}_{3-D,s} \times N_s/(N_l + N_s)$, where $\bar{D}_{3-D,l}$ and $\bar{D}_{3-D,s}$ are the average values of 3-D grain size of large and small grain populations, respectively, and N_l and N_s are the number of large grains and the small grains, respectively.

3. Results and discussion

3.1. Microstructure and grain size distribution

Fig. 2(a)–(d) shows the typical EBSD grain orientation image maps (OIMs) along the sample normal direction with respect to the stereographic triangle for 1, 2, 4 and 20 h annealing, respectively, for the Nb-doped sample, and Fig. 2(e) for 20 h annealing for the undoped sample. In this study, all samples had a Ti-excess (Sr/Ti = 0.996 for undoped and 0.994 for Nb-doped), which should cause abnormal grain growth.²⁶

As shown in Fig. 2(a)–(e), there are two groups of grains, large and small. The 2-D grain size distributions were extracted

from the OIMs and converted to 3-D distributions for all samples. For example, Fig. 1 shows the 3-D histogram from Fig. 2(a) (1 h annealing) and the typical bi-modal distribution. Using the 3-D histogram, quantitative information about the bi-modal distribution can be obtained. For example, at 1 h annealing, the overall grain size *G* is 1.5 μ m, including a small outlying number (0.8%) of larger grains of size 5.4 μ m.

Fig. 3 shows the bi-modal grain sizes of Nb-doped $SrTiO_3$ samples as a function of annealing time. In most cases (1, 2, and 4 h), there is a clear separation between those grains that are small and large in size. These results (bi-modal distribution) are evidence of abnormal grain growth, with grains in the larger size group in 1, 2 and 4 h annealed samples being identified as those that have undergone abnormal grain growth. With increasing annealing time, the fraction of large grains increases as does their overall grain size. By 20 h, all small grains were consumed by



Fig. 2. Typical OIMs showing the orientation and shape of the grains obtained in Nb-doped $SrTiO_3$ samples annealed for (a) 1 h, (b) 2 h (c) 4 h and (d) 20 h and (e) the undoped $SrTiO_3$ sample annealed for 20 h. Samples were annealed at 1425 °C in oxygen. Colour-map in OIMs corresponds to the stereographic triangle of (f) and represents the crystal orientation normal to the specimen surface.



Fig. 3. 3-D grain size of Nb-doped SrTiO₃ samples as a function of annealing time. Samples sintered at 1425 °C in oxygen. The percentage of large grains is given in brackets. Error bars indicate the standard deviation of grain size measurement from different sets measured on three regions that contain at least 600 grains each.

the large grains. It should be noted that although the 20 h sample has a unimodal grain size distribution that would be identified with normal grain growth, from the grain growth evolution it is clear that these grains are from the large grains in the 1, 2, and 4 h annealed samples. So, these grains at 20 h are recognized as large grains in this study.

A comparison between the undoped (Fig. 2(d)) and doped (Fig. 2(e)) samples, sintered under the same conditions and for the same time of 20 h, shows that abnormal grain growth is much faster in the Nb-doped samples than in the undoped samples. The doped sample has an overall grain size $G = 81.9 \,\mu\text{m}$ from 100% large grains compared to $G = 4.0 \,\mu\text{m}$ with 16.7% large grains in the undoped sample. This result is in agreement with earlier studies which show that different grain growth behaviour can be obtained by modifying dopants.²⁷

Fig. 3 illustrates that for the shortest annealing time (1 h), large grains are not frequently observed (only 0.8%), while for the longer annealing time (4 h), 90.6% of grains observed are large. The large grains have a size more than ten times that of the small grains (for 2 and 4 h annealed samples), which implies that the GB velocity for the large grains is larger than that for small grains.

3.2. Grain misorientation texture

From the grain size distribution analysis, all the small grains were consumed by the large grains after a 20 h anneal. The experimental observations of microstructural evolution suggest that some grains grow abnormally faster than other grains. Furthermore, this abnormal grain growth behaviour may also affect grain misorientations. As discussed in the introduction, the question is whether or not preferred Σ 3 grain misorientations exist in polycrystalline SrTiO₃. Therefore, the measurement of textural evolution of grain misorientations through annealing is needed to resolve the Σ 3 disagreement.



Fig. 4. Stereographic projection of the unit triangle showing the percentage of the Mackenzie random misorientation axis distribution falling in eight regions (after Mackenzie³²). The regions (from I to VIII) are separated by solid curves at 5° , 10° , 15° , 30° and 45° from [1 0 0] and at 5° and 9.74° from [1 1 1].

The grain misorientation texture²⁸ is a function of three independent parameters (two for a misorientation axis and one for a misorientation angle). A common strategy for analyzing this 3-D data is to project it into one-dimensional space (misorientation angle)^{29,30} or 2-D space (misorientation axis).^{22,28,31} In this study, the grain misorientations are displayed in misorientation axis and angle space separately to obtain the misorientation axis and angle distributions for texture analysis.

First, to evaluate the distribution of the GB misorientation axes, the experimental data from different annealing times were compared with the theoretical Mackenzie random distribution³²: the probability of a given random misorientation between two cubic crystals. The distribution of the misorientation axis can be displayed in a standard stereographic triangle for the random distribution,³² as shown in Fig. 4. The triangle is divided into eight regions (I-VIII) showing the percentage of the random distribution in each region. The first five regions are divided by the curves at 5°, 10°, 15°, 30° and 45° away from [100]; the other three regions (data with axes orientated at angles greater than 45° from [100]) are divided by lines at 5° and 9.7° from [111]. Fig. 5 shows the GB misorientation distributions of 1 h (with 0.8% large grains) and 20h (with 100.0% large grains) compared to the random distribution, for each of these eight regions. It is seen that the distribution for the 20h sample is more like the random distribution than that of the 1 h sample.

In order to investigate why the GB axis distribution becomes more random for longer annealing times, the regions from the stereographic triangle with the largest deviations from random were analyzed further. In the 1 h sample, 82.4% of the total deviation from the random distribution occurs in four regions: I, IV, V and VIII. Three of these four regions include an area within 5° of a low-index axis: regions I, IV and VIII are near [100], [111] and [101], respectively. This suggests that GB misorientation axes near low-index directions dominate the deviation from randomness. Fig. 6 shows GB misorientation axis frequency near low-index directions as a function of annealing time in Nb-doped SrTiO₃ samples. It is clear that the fraction of the population near the three low-index directions reduces



Fig. 5. Comparison of misorientation axis distribution in Nb-doped SrTiO₃ samples annealed for 1 and 20 h at 1425 $^{\circ}$ C in oxygen with the Mackenzie random distribution. Error bars indicate the standard deviation of grain misorientation frequency from different sets measured on three regions that contain at least 600 grains each.

with increasing annealing time. These data suggest that the distributions of small grains (99.2% in 1 h sample) and large grains (100.0% in 20 h sample) are different: the main difference is from the GB misorientation axes near low-index directions. Two effects are occurring during annealing (a) the large grain population is increasing in number and (b) grain growth is occurring. While it may be tempting to associate the 1 h data with the characteristics of the small grains and the 20 h data with those of the large grains, the effects of competitive grain growth cannot be ignored. However, as abnormal grain growth dominates the samples, the contribution to the GB misorientation axis distribution from near low-angle-axes segments reduces to approach the Mackenzie random distribution.



Fig. 6. Proportion of grain misorientations within 5° of low-index axes [100], [101] and [111] as a function of annealing time in Nb-doped SrTiO₃ samples sintered at 1425 °C in oxygen. Error bars indicate the standard deviation of grain misorientation frequency from different sets measured on three regions that contain at least 600 grains each.



Fig. 7. Comparison of misorientation angle distribution in Nb-doped SrTiO₃ samples sintered for 1 and 20 h at 1425 °C in oxygen with the Mackenzie random distribution. Error bars indicate the standard deviation of grain misorientation frequency from different sets measured on three regions that contain at least 600 grains each.

After analyzing the GB misorientation axis distributions, the GB misorientation angle distributions were compared with the Mackenzie random distribution.³³ Fig. 7 shows the GB misorientation angle distribution in Nb-doped SrTiO₃ samples annealed for 1 h and 20 h together with the Mackenzie random distribution. There are two significant differences between the samples. Firstly, the large fraction of low-angle GBs present in the sample annealed for 1 h reduces in the sample annealed for 20 h. Secondly, the GB misorientation distribution for the 20 h annealed sample is more like the random distribution than is the 1 h sample, which is in agreement with the axis distribution analysis in the previous section. It should be noted that, for the 1 h annealed sample, the main deviation from the random distribution comes from small angle $(3^{\circ} < \theta_m < 15^{\circ})$ (35.9% of the total deviation) and large-angle ($40^{\circ} < \theta_m < 50^{\circ}$) (30.7% of the total deviation) misorientations.

Fig. 8 shows the frequency of GB misorientations of lowangle and large-angle GBs as a function of annealing time in Nb-doped $SrTiO_3$ samples. Populations of low-angle and largeangle GBs at the Mackenzie random distribution are illustrated using dotted and solid lines, respectively. The population of low-angle GB segments decreases and the population of largeangle GB segments increases with increasing annealing time to approach the Mackenzie random populations. The data support the idea that the GB misorientations of low- and large-angle GBs contribute to the reduction in deviation from randomness as abnormal grain growth progresses.

3.3. Σ 3 misorientations

The evolution of grain misorientations can be explained by energy minimization during annealing as follows. During grain growth, energy minimization can be achieved in two ways. (i) The total grain boundary area is decreased. The higher veloc-



Fig. 8. Grain misorientations of low-angle $(3^{\circ} < \theta_m < 15^{\circ})$ and large-angle $(40^{\circ} < \theta_m < 50^{\circ})$ boundaries as a function of annealing time in Nb-doped SrTiO₃ samples sintered at 1425 °C in oxygen. Populations of low-angle and large-angle GBs at the Mackenzie random distribution are illustrated using dotted and solid lines, respectively. Error bars indicate the standard deviation of grain misorientation frequency from different sets measured on three regions that contain at least 600 grains each.

ity GBs (general, high-angle GBs³⁴) will consume the lower velocity GBs. This results in a decrease in the population of low-angle and CSL GBs. (ii) The average grain boundary energy is decreased. The grain boundary character distribution will show an increase in the population of low-energy boundaries, i.e. low-angle and CSL GBs.^{35,36} Seaton and Leach showed this second effect, where the density of Σ 3 grain misorientations increased with annealing time between 0 and 60 min in donor-doped BaTiO₃.^{16,17}

Fig. 9 shows the proportion of Σ 3 GB misorientations from all grains and from small and large grains of Nb-doped SrTiO₃ samples as a function of annealing time. The Σ 3 population



Fig. 9. Proportion of Σ 3 misorientations from all grains and from small and large grains in Nb-doped SrTiO₃ samples as a function of annealing time. The percentage of small grains is given in brackets. Solid line indicates the population of the Σ 3 GBs from the Mackenzie random distribution. Samples were sintered at 1425 °C in oxygen. Error bars indicate the standard deviation of grain misorientation frequency from different sets measured on three regions that contain at least 600 grains each.

decreases with increasing annealing time. According to Randle's analysis using the Mckenzie random distribution,^{33,37} the Σ 3 population for a random distribution of GBs is 1.5%. The result for the Σ 3 population in the 20 h sample (~1.8%) is near Randle's figure, and similar values (~1.8–1.9%) of the Σ 3 populations of the large grains were found in the 2 and 4 h samples. As both low-angle and Σ 3 GB populations decrease with increasing annealing time, the results appear to be dominated by mechanism (i).

We assume that AGG effects on the sample can be removed by examining only the population of small grains. In Fig. 9, the population of $\Sigma 3$ GBs in the small grains appears to be increasing with time (mechanism (ii)), although the error bars are large. Additionally, the grain size barely evolves (1.4–4.2 µm from 1 to 4 h annealing) compared with that of the large grains (5.4–28 µm for 1–4 h annealing). The fraction of small grains decreases significantly with annealing time, see Fig. 3. Small grains leave that population either by shrinking to zero volume or by joining the population of abnormal grains. This may explain the apparently stagnant grain growth, if the largest of the small grains are joining the abnormal grain population. Therefore, the grain growth in the sample as a whole appears to be dominated by mechanism (i).

In the case of undoped SrTiO₃, the sample annealed for 20 h has a lower proportion (1.1%) of large grains than the Nb-doped sample at 20 h (100%) (from Fig. 3). This suggests that the nucleation rate of the abnormal grains is lower in undoped samples than in Nb-doped samples. So, abnormal grains in the undoped sample have not consumed low-energy GBs from small grains to the same extent as in doped samples. Consequently for undoped SrTiO₃, mechanism (i) does not dominate (ii) to the same extent as for Nb-doped SrTiO₃, and Σ 3 misorientations are found at a higher proportion (6.2%) than for doped samples. Thus, in SrTiO₃ the proportion of low-energy GBs (low-angle and CSL GBs) is strongly correlated with the grain growth behaviour, i.e. abnormal grain growth.

We now consider the apparently conflicting results in the literature. Ernst et al.¹⁴ reported that Fe-doped SrTiO₃ had preferred Σ 3 misorientations with five Σ 3 GBs being found for 90 measurements based on Brandon's criterion. However, for undoped SrTiO₃, Saylor et al.¹⁵ found 1276 Σ 3 orientation segments in 75,088 GB segments (1.7%), and concluded that in their sample the proportion of Σ 3 misorientations was near random. They suggested that the result of Ernst et al.¹⁴ was due to TiO₂ excess and Fe-doping. We suggest that this discussion can be resolved by considering abnormal grain growth. From the measurement reported by Ernst et al.,¹⁴ all 90 measured grains are small grains while Saylor's data come from both small (\sim 30–120 µm) and large grains ($\sim 240-530 \,\mu m$))^b. In the population of small grains, the Σ 3 misorientations increase in the population over time; however when abnormal grain growth dominates, the population of Σ 3 misorientations decreases in the overall population with time.

^b Grain size distributions are extracted from the Saylor's Fig. 1(b).

4. Conclusions

The frequency of low-energy GBs (i.e. low-angle and $\Sigma 3$ GBs) observed in polycrystalline SrTiO₃ has been correlated to the annealing conditions. Nb doping increases the rate of grain growth and the nucleation rate of abnormal grains. Consequently the proportion of low-energy GBs is not only determined by annealing conditions but also by the mechanism of abnormal grain growth. The result in this study suggests that abnormal grain growth reduces the population of $\Sigma 3$ GBs in SrTiO₃. An increase in the population of $\Sigma 3$ GBs enhances the ion conductivity of SrTiO₃. Therefore, it is expected that for increased ion conductivity in SrTiO₃, a relatively high population of $\Sigma 3$ GBs can be prepared by retarding abnormal grain growth.

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References

- 1. Minh, N. Q., Ceramic fuel-cells. J. Am. Ceram. Soc., 1993, 76, 563-588.
- Takahash, T. and Iwahara, H., Ionic conduction in perovskite-type oxide solid solution and its application to solid electrolyte fuel cell. *Energy Convers.*, 1971, 11, 105–111.
- Chang, H. Y., Cheng, S. Y., Sheu, C. I. and Wang, Y. H., Core-shell structure of strontium titanate self-grown by a hydrothermal process for use in grain boundary barrier layers. *Nanotechnology*, 2003, 14, 603–608.
- Shen, H., Song, Y. W., Gu, H., Wang, P. C. and Xi, Y. M., A high-permittivity SrTiO₃-based grain boundary barrier layer capacitor material single-fired under low temperature. *Mater. Lett.*, 2002, 56, 802–805.
- Rodrigues, R. P., Chang, H. J., Ellis, D. E. and Dravid, V. P., Electronic structure of pristine and solute-incorporated SrTiO₃. II. Grain-boundary geometry and acceptor doping. *J. Am. Ceram. Soc.*, 1999, 82, 2385–2394.
- Chiang, Y. M., Birnie III, D. P. and Kingery, W. D., *Physical Ceramics—Principles for Ceramic Science and Engineering*. John Wiley & Son, Inc., New York, 1997.
- Kronberg, M. L. and Wilson, F. H., Secondary recrystallization in copper. *T. Am. I. Min. Met. Eng.*, 1949, 185, 501–514.
- Brandon, D. G., Structure of high-angle grain boundaries. Acta Metall., 1966, 14, 1479–1484.
- Ogawa, H., Demura, M., Yamamoto, T. and Sakuma, T., Estimation of PTCR effect in single grain boundary of Nb-doped BaTiO₃. J. Mater. Sci. Lett., 1996, 15, 537–538.
- Hayashi, K., Yamamoto, T. and Sakuma, T., Grain orientation dependence of the PTCR effect in niobium-doped barium titanate. *J. Am. Ceram. Soc.*, 1996, **79**, 1669–1672.
- Kienzle, O., Exner, M. and Ernst, F., Atomistic structure of Sigma = 3, (1 1) grain boundaries in strontium titanate. *Phys. Status Solidi A: Appl. Res.*, 1998, 166, 57–71.
- Ernst, F., Kienzle, O. and Rühle, M., Structure and composition of grain boundaries in ceramics. J. Eur. Ceram. Soc., 1999, 19, 665–673.
- Hutt, S., Kostlmeier, S. and Elsasser, C., Density functional study of the Sigma 3 (111) 1(1)over-0 symmetrical tilt grain boundary in SrTiO₃. J. Phys.Condens. Matter, 2001, 13, 3949–3960.

- Ernst, F., Mulvihill, M. L., Kienzle, O. and Rühle, M., Preferred grain orientation relationships in sintered perovskite ceramics. *J. Am. Ceram. Soc.*, 2001, 84, 1885–1890.
- Saylor, D. M., El Dasher, B., Sano, T. and Rohrer, G. S., Distribution of grain boundaries in SrTiO₃ as a function of five macroscopic parameters. *J. Am. Ceram. Soc.*, 2004, **87**, 670–676.
- Seaton, J. and Leach, C., Evolution of low sigma grain boundaries in PTC thermistors during sintering. J. Eur. Ceram. Soc., 2005, 25, 3055– 3058.
- Seaton, J. and Leach, C., Formation and retention of low Sigma interfaces in PTC thermistors. *Acta Mater.*, 2005, 53, 2751–2758.
- Park, M. B., Shih, S. J. and Cockayne, D. J. H., The preferred CSL misorientation distribution in polycrystalline SrTiO₃. *J. Microsc. Oxford*, 2007, 227, 292–297.
- Bäurer, M., Kungl, H. and Hoffmann, M. J., Influence of Sr/Ti stoichiometry on the densification behaviour of strontium titanate. *J. Am. Ceram. Soc.*, 2009, **92**, 601–606.
- Shih, S. J., Park, M. B. and Cockayne, D. J. H., The interpretation of indexing of high Sigma CSL grain boundaries from ceramics. *J. Microsc. Oxford*, 2007, 227, 309–314.
- Humphreys, F. J., Review—grain and subgrain characterisation by electron backscatter diffraction. J. Mater. Sci., 2001, 36, 3833–3854.
- Lee, J. S., Misorientation distribution of a Mn–Zn ferrite sample with abnormal grain growth. J. Ceram. Process Res., 2004, 5, 179–186.
- Underwood, E. E., *Quantitative Stereology*. Addison-Wesley, Reading, 1970.
- Peng, C. J. and Chiang, Y. M., Grain-growth in donor-doped SrTiO₃. J. Mater. Res., 1990, 5, 1237–1245.
- Bae, C., Park, J. G., Kim, Y. H. and Jeon, H., Abnormal grain growth of niobium-doped strontium titanate ceramics. *J. Am. Ceram. Soc.*, 1998, 81, 3005–3009.
- Chung, S. Y. and Kang, S. J. L., Intergranular amorphous films and dislocations-promoted grain growth in SrTiO₃. *Acta Mater.*, 2003, 51, 2345–2354.
- Chung, S. Y., Yoon, D. Y. and Kang, S. J. L., Effects of donor concentration and oxygen partial pressure on interface morphology and grain growth behavior in SrTiO₃. *Acta Mater.*, 2002, **50**, 3361–3371.
- Randle, V. and Ralph, B., The coincident axial direction (CAD) approach to grain-boundary structure. J. Mater. Sci., 1988, 23, 934–940.
- Bystrzycki, J. and Varin, R. A., Microstructure and microtexture in powderextruded monolithic NiAl and NiAl-HfC alloy. *Intermetallics*, 1998, 6, 277–289.
- Randle, V., Interfacial geometry in simulated rolling and recrystallization textures. *Model. Simul. Mater. Sci.*, 1996, 4, 455–472.
- Lee, J. S., Kim, D. Y., Fleig, J. and Joachim, M., Geometry and electrical properties of grain boundaries in manganese zinc ferrite ceramics. *J. Am. Ceram. Soc.*, 2004, 87, 1895–1902.
- Mackenzie, J. K., Distribution of rotation axes in random aggregate of cubic crystals. *Acta Metall.*, 1964, 12, 223–225.
- Mackenzie, J. K., 2nd paper on statistics associated with the random disorientation of cubes. *Biometrika*, 1958, 45, 229–240.
- Sutton, A. P. and Balluffi, R. W., *Interfaces in Crystalline Materials*. Oxford University Press, Oxford, 1995, pp. 526–573.
- Dimou, G. and Aust, K. T., Relative energies of grain-boundaries near a coincidence orientation relationship in high-purity lead. *Acta Metall.*, 1974, 22, 27–32.
- Randle, V., Application of electron backscatter diffraction to grain boundary characterisation. *Int. Mater. Rev.*, 2004, 49, 1–11.
- Randle, V., Relationship between coincidence site lattice, boundary plane indices, and boundary energy in nickel. *Mater. Sci. Technol.*, 1999, 15, 246–252.